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Measurement of Indoor Radon Concentration in Some Selected Offices of Adigrat University, Tigray Region, Ethiopia

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Abstract: Radon is an inert radioactive gas. A long term exposure to high concentration of radon causes lung cancer. Taking this into account, an investigation in some offices of Adigrat Universityhas been made to estimate the radon concentrations and to determine the health risk of the workers in the community. In the present study, Solid State Nuclear Track Detectors (LR-115 type II) has been used to measure the emission of radon from 12 offices for a period of 3 months. After an exposure time of 3 months, films were removed from each office andtaken for etching. After the etching, films were washed with distilled water and allowed to dry for 1day. Finally, alpha tracks were counted using an optical microscope at magnification power of 400X. Results obtained in the present study show that the value of concentration of radon ranges from 171.31 $Bq.m^{-3}$ to 394.05 $Bq.m^{-3}$ with mean concentration273.79 $Bq.m^{-3}$ and standard deviation of 79.7$Bq.m^{-3}$ and the inhalation dose rate varies from a minimum of 1.54 $mSv.y^{-1}$ to a maximum of 3.55 $mSv.y^{-1}$ with an average of 2.46 $mSv.y^{-1}$ and standard deviation of 0.72 $mSv.y^{-1}$. Even though, radon concentration levels were found to be higher the inhalation dose rate is safe within recommended limits

Keywords: Radon Concentration, LR-115 Plastic Track Detectors, Offices

1. Introduction

The knowledge of the radioactivity level in building materials is important in order to estimate the radiological hazards on human health. It has been found that radon, which is a topic of health concern, is a ubiquitous indoor air pollutant in dwellings to which all persons are exposed. Radon is naturally occurring radioactive noble gas which has no test, odor or color. It produced by the decay of Radium which is foundin rocks, soils and building materials. It comes from the natural decay of uranium 238 series, with atomic number 86 and mass number 222. It was discovered by the German physicist Friedrich Ernst Dorn in 1900, who called it niton. It has been called radon since 1923. Radon condenses to a clear colorless liquid at its boiling point and then freezes to form a yellow, then orange red color solid. Radon is also moderately soluble in water and, therefore, can be absorbed by water flowing through rock and sand containing radon. Its solubility depends on the water temperature, the colder the temperature of water the greater the radon's solubility. Its decay products called radon daughters or radon progeny emit highlyionizing alpha-radiation. Decay products are suspended in the air which we breathe. Although the risk is very low when radon is diluted to extremely low concentrations in the open, radon in room air typically contributes up to 50% to the background radiation. However, in places such as caves and mines, it can accumulate up to dangerous concentrations and may cause substantial health damage after long-term exposure. Radon can also be found in drinking water and this can sometimes present a hazard.

Certain types of geology, such as granite and volcanic soils, as well as aluminous shale's, are more likely to contain radon. Conversely, low concentrations of this gas are expected in sedimentary rocks.

Radon being a gas can migrate by mechanism of diffusion and convection throughpore spaces in the soil, and enter the

atmosphere, thus, reaching the human environment [1-4]. Lung cancer is the principal health concern associated with radon exposure. About half of the total radiation effective dose to the general public isdue to the irradiation of the lungs by alpha particles. Continuous exposure to higher levels of radon leads to malignant transformation which results in lung cancer [5]. The level of exposure to radon in the building depends on ventilation in the building and local geology [6-7]. The studies of radon's behavior in the geological environment have indicated that there is a direct relationship between indoor radon levels and concentration of the gas in soil. Windows and doors which are not air-tight lead to a greater air exchange rateand if the basement or other soil-contacting parts of the building areopenitmakes easy for radon to spread upwards [8]. In present study investigations have beencarried out to measure the radon concentrations in some offices in AdigratUniversity, Ethiopia using Solid State Nuclear Track Detectors (LR-115 type II).

2. Area Under Investigation

Adigrat University is a higher education and training public institution located in the city of Adigrat, situated in the Eastern Tigray Regional State of Ethiopia at longitude and latitude 14°16′N 39°27′ECoordinates with an elevation of 2,457 meters (8,061 ft) above sea level and below a high ridge to the west. It is approximately 900 kilometers north of Addis Ababa, the country's capital [9-10].
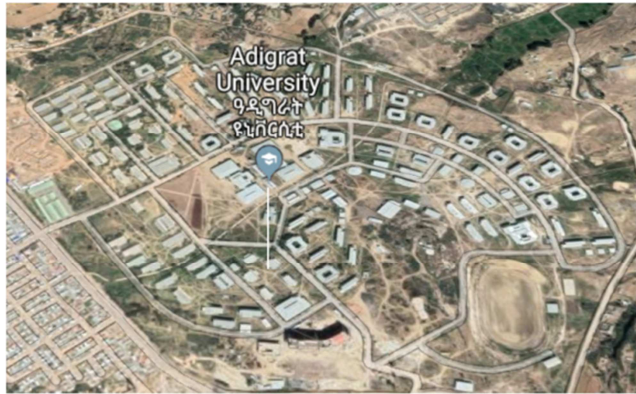

Figure 1. Map of Adigrat University.

3. Experimental Procedure

In the present study, the measurements of radon concentration in offices were carried out by using the LR-115 Type II, plastic track detectors. These plastic film detectors of size 2cm× 3cmwere hanging at 1m away from the roof of offices. Detectors were exposed for a 3 months in ventilated, partially ventilated and unventilated offices. After an exposure time of3 months, films were removed from each officeand taken to Mekelle University for etching. Films were etched with 2.5N NaOH solution at 65°C for 90 minutes. The etching was carried out to reduce the thickness of the LR-115 detectors to about 5 μm. After theetching, films were washed with distilled water and allowed to dry for 1day. Finally, alpha tracks were counted using an optical microscope at magnification of 400x.

4. Results and Discussion

Radon concentration (C_{Rn}) is measured in terms of Bq/m^3, since the concentrations were found by using the following expressions[11-12].

$\rho = \dfrac{N}{A}$, Where ρ is track density, N is number of tracks and A is the area of the detector film (LR-115).

The potential alpha energy concentration Cp(mWL) is given by the formula,

$$C_p(mWL) = \frac{\rho}{KT} \qquad (1)$$

mWL is mili working level which is the unit of Cp, K is the average value of the calibration factor of ^{222}Rn in $(tracks.cm^{-2})/ (days.Bq.m^3)$and T is the exposure time(day).

The concentration of radon and the potential alpha energy concentration is also given by;

$$C_{Rn}(Bq.m^{-3}) = \frac{3.7 C_p}{F} \qquad (2)$$

Where F=0.4 is equilibrium factor.

The inhalation dose D_{in} (in $mSv.y^{-1}$) was calculated using the equation;

$$D_{in} \text{ (in } mSv.y^{-1}) = 0.009 C_{Rn} \qquad (3)$$

Table 1. Observed values of track density, potential alpha energy concentration, radon concentrations and inhalation rates.

Detector code	Track density ρ(tracks.cm⁻²)	Nature ofventilation	CP(MwL)	CRn(Bq.m⁻³)	Din(mSv/y)
D-1	33.33	Excellent	18.52	171.31	1.54
D-2	40	Moderate	22.22	205.54	1.85
D- 3	66.67	Poor	37.04	342.62	3.08
D-4	63.33	Moderate	35.18	325.42	2.93
D-5	36.67	Partial	20.37	188.42	1.70
D-6	55	Moderate	30.55	282.59	2.54
D-7	68.33	Poor	37.96	351.13	3.16
D-8	73.33	Poor	40.74	376.85	3.39
D-9	53.33	Moderate	29.63	274.08	2.47
D-10	35	Excellent	19.44	179.82	1.62
D-11	76.67	Poor	42.60	394.05	3.55
D-12	37.67	Very Good	20.93	193.60	1.74

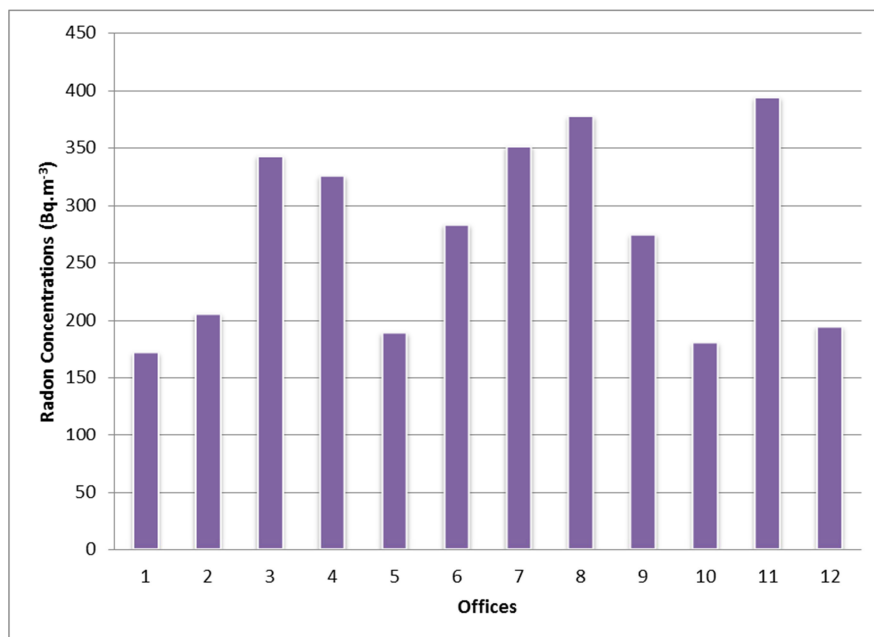| Detector code	Track density ρ(tracks.cm⁻²)	Nature ofventilation	CP(MwL)	CRn(Bq.m⁻³)	Din(mSv/y)
Mean	53.28	_	29.60	273.79	2.46
Standard deviation	15.5	_	8.6	79.7	0.72

Figure 2. *Frequency distribution of radon concentration in offices.*

Figure 2 shows the Frequency distribution of radon concentrations in dwellings of Adigrat University. There is a linear correlation appeared throughout the figure, this finding is reported for the first time from the study area.

Table 1 shows the number of LR-115 detectors, observed values of track density, potential alpha energy concentration, radon concentrations and inhalation rates. Radon concentrations were found to range from a minimum value 171.31 Bq.m⁻³ to a maximum value 394.05 Bq.m⁻³ with mean values of 273.79Bq.m⁻³ and standard deviation of 79.7Bq.m⁻³. Inhalation dose rate varies from a minimum of 1.54 mSv.y⁻¹ to a maximum of 3.55 mSv.y⁻¹ with an average of 2.46 mSv.y⁻¹ and standard deviation of 0.72 mSv.y⁻¹· The obtained values of radon concentration are higher than the recommended limits whereas, inhalation dose rate is within the safe limits. From the result observed the variation of radon concentration is high. The high radon level in some offices suggests that the difference in air dilution, ventilation system, building materials used (like mud and cements) and mood of construction of offices influence the indoor radon levels. Occupants in that area are expected to expose to a high level of doses with a potential risk of lung cancer. After completion of the countrywide radon assessments a detail comparison of results will be available.

5. Conclusion

Radon concentrations were found to range from a minimum value 171.31 Bq.m⁻³ to a maximum value 394.05 Bq.m⁻³ with mean values of 273.79Bq.m⁻³ and standard deviation of 79.7Bq.m⁻³. Inhalation dose rate varies from a

minimum of 1.54 mSv.y⁻¹ to a maximum of 3.55 mSv.y⁻¹ with an average of 2.46 mSv.y⁻¹ and standard deviation of 0.72 mSv.y⁻¹· The obtained values of radon concentration are higher than the recommended limits whereas, inhalation dose rate is within the safe limits. From the result observed the variation of radon concentration is high. From the results found it may be concluded that the observed values of radon concentrations in the offices of Adigrat university arehigher while, annual effective dose are within the safe limit as prescribed by InternationalCommission on Radiological Protection (ICRP) recommends an action level of 200 Bq.m⁻³ forradon and annual effective dose of 3-10 mSv/y [13].

Acknowledgements

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